

AMENDMENTS TO THE CLAIMS

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

1. (Currently Amended) A pyroelectric compound, comprising:

a metal, a mixture of metals, or a semi conducting compound[[,]] lacking spatial periodicity;

said pyroelectric compound being an inorganic, quasi-amorphous oxide compound having piezoelectric properties;

said pyroelectric compound being a product of application of a mechanical strain to a substantially amorphous compound, said mechanical strain being controlled so as to prevent crystallization of said compound. wherein said pyroelectric compound is an inorganic, quasi-amorphous oxide compound having pyroelectric properties.

2. (Withdrawn) The compound of claim 1 having the formula $(AxBy_{1-x})_pO_n$, wherein A and B are independently selected from transition metals, elements of Group IVA of the periodic table, alkali metals, alkali earth metals and rare earth metals; x

has values of between 0 to 1; p is an integer having the values 1, 2 or 3; and n is an integer having the value of 1, 2, 3 or 4.

3. **(Withdrawn)** The compound of claim 2, wherein A is a transition metal or an element of Group IVA of the periodic table, x is 1 and p is 2.

4. **(Original)** The compound of claim 1, having the formula $(A_xB_{1-x})(C_yD_{1-y})O_n$ wherein A and B are independently selected from alkali metals, alkali earth metals, rare earth metals and elements of Group IVA of the periodic table; C and D are independently selected from transition metals and alkali earth metals; x and y have values of between 0 to 1; and n is an integer having the value of 1, 2 or 3.

5. **(Original)** The compound of claim 4, wherein A and B are independently selected from Ba, Sr, Ca, Pb, La, Eu, Li, Na, K and Cs ; C and D are independently selected from Ti, Zr, Nb, Ta, Sc, Mg and V; and n is 3.

6. **(Original)** The compound of claim 5, wherein A and B are independently selected from Ba, Sr, Ca, Pb, La and Eu.

7. (**Original**) The compound of claim 5, wherein A and B are independently selected from Li, Na, K and Cs.

8. (**Original**) The compound of claim 5, wherein C and D are independently selected from Ti and Zr.

9. (**Original**) The compound of claim 6, wherein C and D are independently selected from Ti and Zr.

10. (**Original**) The compound of claim 7, wherein C and D are independently selected from Ti and Zr.

11. (**Original**) The compound of claim 5, wherein C and D are independently selected from Nb, Ta, Sc, Mg and V.

12. (**Original**) The compound of claim 6, wherein C and D are independently selected from Nb, Ta and V.

13. (**Original**) The compound of claim 7, wherein C and D are independently selected from Nb, Ta and V.

14. **(Original)** Inorganic, quasi-amorphous compound of claim 4, wherein $y=0$ and having the formula $(A_xB_{1-x})DO_3$, wherein A, B, D and x are as defined in claim 4.

15. **(Original)** The compound of claim 4 having a pyroelectric coefficient of between about 10^{-12} C/(cm² x K) and about 10^{-7} C/(cm² x K).

16. **(Original)** The compound of claim 14 having a pyroelectric coefficient of between about 10^{-12} C/(cm² x K) and about 10^{-7} C/(cm² x K).

17. **(Original)** The compound of claim 4 selected from BaTiO₃, CaTiO₃, PbTiO₃, Pb(ZrTi)O₃, Pb(Zr_{0.35}Ti_{0.65})O₃, (PbCa)TiO₃, (PbLa)(ZrTi)O₃, PbLaTiO₃, Pb(ScTa)O₃, Pb(ScNb)O₃, Pb(MgNb)O₃, SrTiO₃, (Sr_{0.65},Ba_{0.35})TiO₃, (Ba_{0.70},Sr_{0.30})TiO₃ and EuTiO₃.

18. **(Original)** The compound of claim 17 having a pyroelectric coefficient of between about 10^{-12} C/(cm² x K) and about 10^{-7} C/(cm² x K).

19. **(Original)** The compound of claim 17 being selected from BaTiO_3 , PbTiO_3 and SrTiO_3 .

20. **(Original)** The compound of claim 18 being BaTiO_3 .

Claims 21-23. **(Cancelled)**

24. **(Currently Amended)** Inorganic quasi-amorphous compound of the formula $(\text{AxB}_{1-x})(\text{CyD}_{1-y})\text{O}_3$,

wherein A and B are independently selected from alkali metals, alkali earth metals, rare earth metals and elements of Group IVA of the periodic table;

C and D are independently selected from transition metals and alkali earth metals;

x and y have values of between 0 to 1;

lacking spatial periodicity; and

~~preparable by~~ wherein said compound is a product of
applying a mechanical strain to a substantially amorphous compound of the formula $(\text{AxB}_{1-x})(\text{CyD}_{1-y})\text{O}_n$ wherein n is an integer having the value of 1, 2 or 3, said mechanical strain being controlled so as to prevent crystallization of said compound, thereby obtaining

inorganic quasi-amorphous compound having pyroelectric properties..

25. **(Original)** A device comprising the compound according to claim 1 in the form of a film coating on a substrate.

26. **(Original)** A device comprising the compound according to claim 4 in the form of a film coating on a substrate.

27. **(Original)** The device of claim 26, wherein the substrate is selected from Si, SiO₂ and glass.

28. **(Original)** The device of claim 27, wherein the thickness of the coating layer is below 0.5 micron.

29. **(Previously Presented)** A device comprising the compound of claim 1, the device being operable as a sensor for sensing an external field including at least one of the following: temperature field, magnetic field and electric field.

30. **(Previously Presented)** A device comprising the compound of claim 4, the device being operable as a sensor for

In re Appl of Igor Lubomirsky
Appln. No. 10/560,107
Amendment dated November 20, 2009
Reply to Office Action mailed May 20, 2009

sensing an external field including at least one of the following:
temperature field, magnetic field and electric field.

31. **(Original)** A device having an acoustic wave propagation element including the compound of claim 1.

32. **(Original)** A device having an acoustic wave propagation element including the compound of claim 4.

33. **(Original)** A device having an acoustic wave propagation element including the compound of claim 5.

34. **(Original)** A birefringent medium comprising the compound of claim 1.

35. **(Original)** A birefringent medium comprising the compound of claim 4.

36. **(Original)** A device comprising the compound according to claim 1.

In re Appl of Igor Lubomirsky
Appln. No. 10/560,107
Amendment dated November 20, 2009
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37. **(Original)** A device comprising the compound according to claim 4.

38. **(Withdrawn)** A device comprising a compound according to claim 3 in the form of a film coating on a substrate.

39. **(Withdrawn)** The device of claim 38, wherein the substrate is selected from Si, SiO₂ and glass.

40. **(Withdrawn)** The device of claim 39, wherein the compound is SiO₂.

41. **(Currently Amended)** The pyroelectric compound of claim 1, which is a non-crystalline ionic solid having macroscopic polarization and a lack of spatial periodicity.